

Subscriber access provided by ISTANBUL TEKNIK UNIV

Pigments of Fungi, Part 27. New Xanthorin **Derivatives from a Fungus of the Genus Dermocybe**

Melvyn Gill, Asfia Qureshi, and Roy Watling

J. Nat. Prod., 1992, 55 (4), 517-520• DOI: 10.1021/np50082a023 • Publication Date (Web): 01 July 2004

Downloaded from http://pubs.acs.org on April 4, 2009

More About This Article

The permalink http://dx.doi.org/10.1021/np50082a023 provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

PIGMENTS OF FUNGI, PART 27.¹ NEW XANTHORIN DERIVATIVES FROM A FUNGUS OF THE GENUS DERMOCYBE

MELVYN GILL,* ASFIA QURESHI,

School of Chemistry, The University of Melbourne, Parkville, Victoria 3052, Australia

and ROY WATLING

Royal Botanic Garden, Edinburgh, Scotland EH3 5LR

ABSTRACT.—The hydroxylated anthraquinones 2 and 4 were isolated from an indigenous Australian mushroom of the genus *Dermocybe*, and their structures were established by spectroscopic and chemical methods.

Hydroxylated anthraquinones occur in great variety in the fruiting bodies of fungi belonging to the closely related genera Cortinarius and Dermocybe. [The rank of Dermocybe remains a matter for debate. In line with most recent chemical publications (1), and consistent with Moser (2), we have elected to treat Dermocybe as a genus. Other authorities regard Dermocybe as a sub-genus of Cortinarius.] More than thirty pigments of this type are now known from European, North American, and Scandinavian species where their presence has assumed considerable taxonomic importance (1). To date, comparatively little study has been made of Dermocybe species indigenous to the Southern Hemisphere; nevertheless, there are now clear indications that the secondary metabolism of Australasian dermocybes is sufficiently different from that of their Northern



¹For Part 26, see M. Gill and A. Giménez, J. Nat. Prod., in press.

Hemisphere counterparts to warrant systematic chemical study (3–7).

We report here the isolation from a diminutive, yellow-brown Australian *Dermocybe* mushroom [WAT 22963 (Cortinariaceae)] of two anthraquinone pigments and describe their identification as 2 and 4, two new derivatives of the known natural product xanthorin (1,5,8-trihydroxy-6-methoxy-3-methyl-9, 10-anthracenedione) [1] (8-10).

RESULTS AND DISCUSSION

Fresh fruiting bodies of WAT 22963 were chopped and macerated in EtOH overnight. Subsequent operations served to concentrate the extracts, distribute the pigments between EtOAc and H_2O , and concentrate the coloring matters in the organic phase. Preparative tlc then afforded two red pigments, each in pure form.

The chromatographically more mobile pigment (57 ppm fresh wt) was obtained as red needles to which the molecular formula $C_{17}H_{14}O_6$ was assigned by hreims ($[M]^+ m/z 314.0791$). The electronic spectrum, with absorption maxima at 467, 496, and 532 nm, was suggestive of a 1,4-dihydroxylated anthraquinone (11). The ¹H-nmr spectrum revealed the presence of two chelated hydroxyls (δ 13.43 and 13.98), two OMe groups (δ 3.99 and 4.07), an isolated aromatic proton (δ 6.71), two meta-coupled (J = 0.8 Hz) aromatic protons (δ 7.17 and 7.86), and an aromatic C-Me group (δ 2.53). These data are consistent with, but do not uniquely define, the anthraquinone structure 2 for this red pigment.

The arrangement of the substituents in each individual peripheral ring in the anthraquinone nucleus of 2 was established by differential nOe experiments. Thus, irradiation of the protons of the higher field OMe group (δ 3.99) caused enhancement (8%) of only the isolated aromatic proton resonance (δ 6.71). On the other hand, irradiation of the protons of the alternative OMe group (δ 4.07) caused enhancement (6%) in the signal (δ 7.17) from one of the two metacoupled aromatic protons. This result places one OMe together with the isolated aromatic proton and both chelated OH groups in the same peripheral ring, with the second OMe, the meta-coupled protons, and the C-Me group in the other. In accord with this conclusion, irradiation of the C-Me group (δ 2.53) enhanced the signals from both of the meta-coupled aromatic protons.

The connectivity between the two ends of the anthraquinone 2 was established unequivocally by direct chemical correlation with the tri-O-methyl ether 3 of xanthorin [1] isolated from Dermocybe splendida (10). Treatment of the red pigment 2 with Me₂SO₄ and K₂CO₃ in Me₂CO gave the corresponding di-Omethyl ether, $C_{19}H_{18}O_6$, which was indistinguishable chromatographically and spectroscopically from the quinone 3 obtained from xanthorin [1] by permethylation under the same conditions. Xanthorin-1-O-methyl ether [2] is a new natural product.

The more polar red pigment (86 ppm fresh wt) was obtained as red needles, $C_{17}H_{14}O_7$, $[M]^+ m/z$ 330.0737, containing one oxygen atom more than its co-metabolite **2**. The electronic spectrum ($\lambda \max 467, 495$, and 531 nm) suggested that this second anthraquinone possesses the same chromophore as **2**, which is in full accord with the ¹H-nmr data. Thus, singlets at δ 13.96, 13.41, 6.72, 4.10, and 3.99 reveal the presence of two chelated hydroxyls, an isolated aromatic proton, and two OMe groups, respectively, while the appearance of a pair of meta-coupled doublets (J = 0.8 Hz) at δ 7.48 and 7.97 shows that the extra oxygen atom is not attached directly to the anthraquinone nucleus. Its location as an OH substituent on the 3-Me group is evident by the replacement of the three-proton singlet (δ 2.53) observed in the spectrum of **2** with a two-proton signal at δ 4.89 in the spectrum of its more polar counterpart. The formula **4**, therefore, emerges for this new natural product.

The arrangement of the substituents in each individual peripheral ring of the anthraquinone nucleus in 4 was verified by nOe experiments that parallelled those performed previously on xanthorin-1-0-methyl ether [2].

The correct connectivity between the two ends of the anthraquinone 4 was proved by chemical correlation with the (new) permethyl ether 5 of the known insect pigment 6 (12). Accordingly, an authentic sample of ω , 5-dihydroxyemodin [6] isolated from the ericoccid *Eriococcus coriaceus* (12) was methylated by using Me₂SO₄/K₂CO₃ in Me₂CO. The methyl ether 5, C₁₉H₁₈O₇, so obtained was identical in all respects with the di-O-methyl ether of the new natural product 4.

 ω -Hydroxyxanthorin-1-O-methyl ether [4] is only the second example from within the large group of *Cortinarius* and *Dermocybe* pigments of an anthraquinone bearing an ω -OH group. The other is fallacinol, which occurs along with its ω -O-acetyl derivative in *Dermocybe cinnabarina* (13).

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.— ¹H-nmr spectra were measured at 399.65 MHz on a JEOL JNM-GX 400 spectrophotometer for solutions in CDCl₃. Ei (70 eV) mass spectra were obtained wth a V.G. Micromass 7070F spectrometer. Uv spectra were recorded with a Varian SuperScan 3 spectrophotometer for solutions in EtOH. Preparative tlc was performed on layers $(20 \times 20 \times 0.1 \text{ cm})$ of Merck Kieselgel GF₂₅₄ Si gel with toluene-HCO₂Et-HCO₂H (50:49:1) as eluent.

FUNGUS MATERIAL.—Mushrooms were collected in July 1990 from mixed *Eucalyptus* forest on the western slopes of Mount Slide in the Kinglake National Park, Victoria, Australia. A voucher specimen is lodged in the herbarium of the Royal Botanic Garden, Edinburgh, UK, under accession number WAT 22963.

TAXONOMIC DIAGNOSIS OF WAT 22963.-Basidia 4-spored, $4.8-5.2 \times 17.5-22 \ \mu m \ excl.$ sterigmata (1.3-1.8 mm long). Clavate with basal pedicel, hyaline becoming honey-yellow with age. Basidiospores 6-6.5 \times (3.5)-4 μ m elliptic sometimes broadly so or very slightly amygdaliform in side-view, elliptic in face-view with broad apex, verruculose to strongly verrucose at apex and appearing as if forming a cap; pale golden brown in H₂O, slightly darker in ammoniacal solutions, relatively thick-walled, germ-pore absent, small but distinct hyaline apiculus present. Cheilocystidia forming an irregular margin 11- 15×5.5 -6.5 µm, irregularly clavate, bluntly fusoid to elongate vesiculose, pale to strongly pigmented tawny orange; pleurocystidia absent. Hymenophoral trama regular, of broad to very broad hyphae ranging from 12-14 µm in mediostratum to 20-25 µm in lateral stratum delimited by a narrow zone of parallel hyphae $3-4 \,\mu\text{m}$ broad immediately beneath the hymenial/subhymenial cells; hymenium with dispersed amorphous collections of rich orange-brown to rich tawny material in ammoniacal solutions. Pileipellis a cutis 5-6 cells deep of radially arranged broadly ellipsoid to shortly cylindric, branched, tawny, relatively thick-walled, generally smooth hyphae 28- $33 \times (11)$ -22 µm broad with membranal pigmentation; velar cells on pileipellis of purplish date, branched, clamp-connected hypl.ae 6.5-8.7 µm broad; pilocystidia absent. Context of similar cells to pileus or more variable and slightly longer, less pigmented cells with paler membranal pigment, differentiated at top by smaller ellipsoid cells flattened in a radial plane, and at base with filamentous hyphae 3-4 µm broad supporting hymenium. Stipitipellis with an outer layer of filamentous, red-brown, flexuous hyphae 7.5-9 µm broad often adhering together overlying a layer of less pigmented, more regular, cylindric cells; outer layer with velar hyphae similar in all respects to those on pileus. Clamp-connections throughout.

The species is placed close to Dermocybe vinicolor Horak (15).

EXTRACTION AND ISOLATION.—Fresh mushrooms (7 g) were chopped and soaked in EtOH (150 ml) at room temperature for 12 h. The extract was evaporated, and the residue was partitioned between EtOAc (3×50 ml) and H₂O (50 ml). The dried (Na₂SO₄) organic layers were evaporated, and the residue (26 mg) was purified by preparative tlc to afford the anthraquinones **2** (0.4 mg, 57 ppm fresh wt, R_f 0.55) and **4** (0.6 mg, 86 ppm fresh wt, R_f 0.20).

Xanthorin-1-O-methyl ether [2].—Red needles (CHCl₃, -20°): mp 205–209°; found [M]⁺ m/z314.0791 (C₁₇H₁₄O₇ requires 314.0790); uv λ max (log ϵ) 221 (2.81), 230 (2.77), 236 (2.77), 256 (2.84), 294 (2.40), 467 (2.34), 496 (2.40), 532 (2.20) nm; ¹H nmr δ 2.53 (s, 3-Me), 3.99 (s, 6-OMe), 4.07 (s, 1-OMe), 6.71 (s, H-7), 7.17 (d, J = 0.8 Hz, H-2), 7.86 (d, J = 0.8 Hz, H-4), 13.43 and 13.98 (each s, 5-OH, 8-OH); ms m/z[M]⁺ 314 (100), 296 (48), 268 (29), 253 (22), 225 (12), 115 (14), 77 (11), 15 (12).

ω-Hydroxyxanthorin-1-O-methyl ether [4].— Red needles (CHCl₃, -20°): mp 220-223°; found [M]⁺ m/z 330.0737 (C₁₇H₁₄O₇ requires 330.0739); uv λ max (log ε) 222 (2.75), 230 (2.72), 236 (2.73), 257 (2.82), 293 (2.39), 467 (2.31), 495 (2.36), 531 (2.16) nm; ¹H nmr δ 3.99 (s, 6-OMe), 4.10 (s, 1-OMe), 4.89 (d, J= 4.4 Hz, 3-CH₂), 6.72 (s, H-7), 7.48 (d, J= 0.8 Hz, H-2), 7.97 (d, J= 0.8 Hz, H-4), 13.41 and 13.96 (each s, 5-OH, 8-OH), ω-OH obscured; ms m/z [M]⁺ 330 (77), 312 (48), 284 (33), 269 (21), 241 (14), 115 (14), 77 (15).

1,5,6,8-Tetramethoxy-3-methyl-9,10-ANTHRACENEDIONE [3].—From xanthorin [1].— A mixture of xanthorin (1 mg), anhydrous K₂CO₃ (200 mg), and Me₂SO₄ (1 drop) in dry Me₂CO (1 ml) was heated under reflux for 4 h. The reaction was cooled, the products were partitioned between EtOAc (20 ml) and H_2O (6 × 20 ml), and the colorless aqueous phase was discarded. The organic phase was dried (Na2SO4) and evaporated, and the residue in MeOH was filtered through a column containing Sephadex LH-20. Crystallization from EtOAc/hexane gave the anthraquinone 3 (1 mg) as yellow needles: mp 188-189° [lit. (14) $185-186^{\circ}$; found $[M]^+ m/z 342.1104$ $(C_{19}H_{18}O_6 \text{ requires } 342.1103); \text{ uv } \lambda \max(\log \epsilon)$ 226 (2.94), 256 (2.76), 280 (2.65), 406 (2.27) nm; ¹H nmr δ 2.44 (s, 3-Me), 3.93 (s, OMe), 3.96 (s, 3 × OMe), 6.77 (s, H-7), 7.02 (br s, H-2), 7.51 (br s, H-4); ms m/z [M]⁺ 342 (100), 327 (98).

From xanthorin-1-O-methyl ether [2].—A mixture of 2 (0.4 mg), anhydrous K_2CO_3 (200 mg) and Me_2SO_4 (1 drop) in dry Me_2CO (1 ml) was heated under reflux for 4 h, and the mixture was worked up and purified as before. Gel permeation gave the anthraquinone 3 (0.4 mg) identical with material described above.

ω-Hydroxy-1,5,6,8-tetramethoxy-3-METHYL-9, 10-ANTHRACENEDIONE [5]. ---From ω, 5-dihydroxyemodin [6].—Compound 6 (5 mg) was converted to 5 under the same conditions used to obtain 3 from 1. The reaction was cooled, and the products were distributed between CH_2Cl_2 (20 ml) and H_2O (6 × 20 ml). The organic phase was dried (Na2SO4) and evaporated, and the residue was purified by preparative tlc to afford the anthraquinone 5 (1.7 mg) as yellow needles (EtOAc/hexane): mp 157-158° (dec); found [M]⁺ m/z 358.1052 (C₁₉H₁₈O₇ requires 358.1052); ir ν max 1666 cm⁻¹; uv λ max (log ϵ) 228 (2.64), 256 (2.45), 281 (2.40), 410 (3.06) nm; ¹H nmr δ 3.83 (s, OMe), 3.94 (s, 3 × OMe), 5.22 (s, 3-CH₂), 6.78 (s, H-7), 7.70 (br s, H-2), 7.90 (br s, H-4); ms m/z [M]⁺ 358 (11), 327 (100).

From ω -hydroxyxantborin-1-O-methyl ether [4]. Anthraquinone 4 (0.6 mg) was converted to 5 under the same conditions used to obtain 5 from 6, and the mixture was worked up and purified as before. Preparative tlc gave the anthraquinone 5 (0.4 mg) identical with material described above.

ACKNOWLEDGMENTS

We thank Professor D. W. Cameron, The University of Melbourne, for an authentic sample of ω , 5-dihydroxyemodin, and Mr. T. May, National Herbarium of Victoria, for help in the location and collection of fungi. Generous financial support was received from the Australian Research Council.

LITERATURE CITED

- M. Gill and W. Steglich, Fortschr. Chem. Org. Naturst., 51, 1 (1987).
- M. Moser, Schweiz. Z. Pilzkd., 50, 153 (1972).
- M. Gill and A. Giménez, J. Chem. Soc., Perkin Trans. 1, 1159 (1990).
- M. Gill, A.F. Smrdel, R.J. Strauch, and M.J. Begley, J. Chem. Soc., Perkin Trans. 1, 1583 (1990).
- C. J. Burns, M. Gill, and A. Giménez, Tetrahedron Lett., 30, 7269 (1989).
- M. Gill and A. Giménez, J. Chem. Soc., Perkin Trans. 1, 2585 (1990).
- M. Gill and A. Giménez, *Phytochemistry*, 30, 951 (1991).
- W. Steglich, W. Losel, and W. Reininger, Tetrabedron Lett., 4719 (1967).
- 9. J. Santesson, *Phytochemistry*, 9, 2149 (1970).
- 10. M.A. Archard, M. Gill, and R.J. Strauch, *Phytochemistry*, 24, 2755 (1985).
- R.H. Thomson, "Naturally Occurring Quinones," 2nd ed., Academic Press, London, 1971.
- H.J. Banks, D.W. Cameron, and M.J. Crossley, Aust. J. Chem., 29, 2231 (1976).
- W. Steglich and W. Reininger, Chem. Ber., 105, 2922 (1972).
- P. Brassard and J.L. Grandmaison, J. Org. Chem., 43, 1435 (1978).
- 15. E. Horak, Sydowia, 40, 81 (1987).

Received 23 July 1991